# DUAL CLUMPED ISOTOPES ( $\Delta_{47}$ and $\Delta_{48}$ ) of Modern Carbonate Sediments

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## **PROJECT OBJECTIVES**

- Measure the  $\Delta_{47}$  and  $\Delta_{48}$  values in modern biogenic and abiogenic sediments.
- Using the dual clumped proxy to ascertain the degree of equilibrium in Modern carbonates.
- Understanding the mechanisms of sediment formation (e.g. direct precipitation or breakdown from existing allochems).

## **PROJECT RATIONALE**

The application of clumped isotopes has caused a revolution in the Earth Sciences. By examining the difference between the measured and the theoretical 47/44 mass ratios, the calculated value ( $\Delta_{47}$ ) has been found to be directly related to the temperature of carbonate formation/alteration (Wang et al., 2004; Ghosh et al.,

2006). However, some carbonates, such corals as (Thiagarajan al., et 2011; Saenger et al., 2012), cave calcites (Daeron et al., 2011; Affek et al., 2014) and microbial associated dolomites (Murrav et al., 2021), show unrealistic  $\Delta_{47}$ temperatures attributable to non-equilibrium processes during their formation. Recently, it has been possible to also measure the  $\Delta_{48}$  value and when used together with the  $\Delta_{47}$  value, these two clumped species are known as the dual clumped isotope proxy. This coupled proxy is able to discern the kinetic isotope effects (e.g. CO<sub>2</sub> absorption and degassing) present in corals



Figure 1: The crossplot of  $\Delta_{47}$  and  $\Delta_{48}$  values and differential kinetic behaviors in different carbonates.

(Bajnai et al., 2020), cave calcites (Fiebig et al., 2021), spring carbonates (Parvez et al., 2023), and microbial-induced dolomites (Lu and Swart, 2023) (Fig. 1). In this study we use the dual clumped proxy to better understand the formation of sediments on the Great Bahama Bank (GBB), the origin of which has long been controversial (Broecker and Takahashi, 1966; Shinn et al., 1989; Morse et al., 2003; Swart et al., 2014; Purkis et al., 2019; Purkis et al., 2023). The samples used in this study were collected by Swart et al (2008) and Reijmer et al. (2009) and represent a range of compositions from mudstone to grainstone.

#### **PRELIMINARY DATA**

The preliminary clumped isotope data shows that while there are no statistically significant differences in the  $\Delta_{47}$  values between the different facies, the  $\Delta_{48}$  values are significantly elevated in the mud dominated facies. (Fig. 2A and B). When comparing the  $\Delta_{47}$  and  $\Delta_{48}$  values to the calibration line of Swart et al. (2021), most of the muddy sediments (mudstone and wackestone) fall below the calibration line (Fig. 3C). In contrast, the grainy sediments (packstone and grainstone) appear to have a bimodal distribution, with values falling not only below the calibration, but also above (Fig. 3C). The possible reason for those variations of  $\Delta_{47}$  and  $\Delta_{48}$  values can be explained by multiple sources of sediments and the different disequilibrium formation processes by which these components were formed.



Figure 2: The crossplot of  $\Delta_{47}$  and  $\Delta_{48}$  values depending on facies in the GBB; a facies number of 1 corresponds to a mudstone, while a facies number of 5 is a grainstone; (A) The  $\Delta_{47}$  values in different facies; (B) The  $\Delta_{48}$  values in different facies; (C) The  $\Delta_{47}$  and  $\Delta_{48}$ values in different facies compare with the calibration line. From Facies 1.5 to 4, Modern sediments become grainier.

The spatial distribution of the  $\Delta_{47}$  and  $\Delta_{48}$  values show the heterogeneous  $\Delta_{47}$  temperatures across the platform top. The  $\Delta_{47}$  temperatures of the muds show lower  $\Delta_{47}$  temperatures (<25°C) than those (>25°C) in the western margin (Fig. 3A). In addition, the area in which the whitings are found show a significant higher  $\Delta_{48}$  values than the marginal area (Fig. 3B). This indicates that the intense precipitation associated with disequilibrium processes occurs in the intraplatform in which the mud accumulates.

#### **PROPOSAL WORK**

In order to further investigate the kinetic effects of dual clumped isotopes in modern sediments on GBB, we propose to; 1) Continue to measure  $\Delta 47$  and  $\Delta 48$  values of surface sediments of all the ~ 260 samples collected in triplicates; 2) Some preliminary measurements have been made on  $\Delta 47$  and  $\Delta 48$  values of red and green algae and these values appear to plot on the calibration line proposed by Swart et al. (2021). However, more species, replicates, and locations need to be measured and compared with the modern sediments; 3) Evaluate the influence of contamination of the  $\Delta_{47}$  and  $\Delta_{48}$  values. The  $\Delta 48$  value had been previously used as a measure of

contamination of the  $\Delta_{47}$  value of a sample. It is important to make sure that, the  $\Delta_{48}$  values that are measured, particularly values that are more positive than expected, are not artifacts.

## SIGNIFICANCE

This study will establish base line values for the  $\Delta_{47}$  and  $\Delta_{48}$  values of Modern sediments as well as casting light on the origin of some of the more controversial sediments, such as the muds in the GBB. Preliminary data suggest that all the surface samples on GBB have formed in disequilibrium with respect to the  $\Delta_{47}$  and  $\Delta_{48}$  values, a finding that has profound implications on the use of these proxies in the geological record.



Figure 3: The distribution of  $\Delta_{47}$  temperatures and  $\Delta_{48}$  values in the GBB.

## REFERENCES

- Affek, H. P., Matthews, A., Ayalon, A., Bar-Matthews, M., Burstyn, Y., Zaarur, S., and Zilberman, T., 2014, Accounting for kinetic isotope effects in Soreq Cave (Israel) speleothems: Geochimica Et Cosmochimica Acta, v. 143, p. 303-318.
- Bajnai, D., Guo, W., Spotl, C., Coplen, T. B., Methner, K., Loffler, N., Krsnik, E., Gischler, E., Hansen, M., Henkel, D., Price, G. D., Raddatz, J., Scholz, D., and Fiebig, J., 2020, Dual clumped isotope thermometry resolves kinetic biases in carbonate formation temperatures: Nature Communications, v. 11, no. 1, p. 4005.

- Broecker, W. S., and Takahashi, T., 1966, Calcium carbonate precipitation on the Bahama Banks: Journal of Geophysical Research, v. 71, no. 6, p. 1575-1602.
- Daeron, M., Guo, W., Eiler, J., Genty, D., Blamart, D., Boch, R., Drysdale, R., Maire, R., Wainer, K., and Zanchetta, G., 2011, <sup>13</sup>C<sup>18</sup>O clumping in speleothems: Observations from natural caves and precipitation experiments: Geochimica et Cosmochimica Acta, v. 75, no. 12, p. 3303-3317.
- Fiebig, J., Daëron, M., Bernecker, M., Guo, W., Schneider, G., Boch, R., Bernasconi, S. M., Jautzy, J., and Dietzel, M., 2021, Calibration of the dual clumped isotope thermometer for carbonates: Geochimica et Cosmochimica Acta, v. 312, p. 235-256.
- Ghosh, P., Adkins, J., Affek, H., Balta, B., Guo, W., Schauble, E. A., Schrag, D., and Eiler, J. M., 2006, <sup>13</sup>C–<sup>18</sup>O bonds in carbonate minerals: A new kind of paleothermometer: Geochimica et Cosmochimica Acta, v. 70, no. 6, p. 1439-1456.
- Lu, C., and Swart, P. K., 2023, The application of dual clumped isotope thermometer ( $\Delta_{47}$  and  $\Delta_{48}$ ) to the understanding of dolomite formation: Geology, v. 52, no. 1, p. 56-60.
- Morse, J. W., Gledhill, D. K., and Millero, F. J., 2003, CaCO<sub>3</sub> precipitation kinetics in waters from the great Bahama bank: Implications for the relationship between bank hydrochemistry and whitings: Geochimica et Cosmochimica Acta, v. 67, no. 15, p. 2819-2826.
- Murray, S. T., Higgins, J. A., Holmden, C., Lu, C., and Swart, P. K., 2021, Geochemical fingerprints of dolomitization in Bahamian carbonates: Evidence from sulphur, calcium, magnesium and clumped isotopes: Sedimentology, v. 68, no. 1, p. 1-29.
- Parvez, Z. A., Lucarelli, J. K., Matamoros, I. W., Rubi, J., Miguel, K., Elliott, B., Flores, R., Ulrich, R. N., Eagle, R. A., Watkins, J. M., Christensen, J. N., and Tripati, A., 2023, Dual carbonate clumped isotopes (Δ<sub>47</sub>-Δ<sub>48</sub>) constrains kinetic effects and timescales in peridotite-associated springs at The Cedars, Northern California: Geochimica et Cosmochimica Acta, v. 358, p. 77-92.
- Purkis, S. J., Harris, P., and Cavalcante, G., 2019, Controls of depositional facies patterns on a modern carbonate platform: Insight from hydrodynamic modeling, v. 5, no. 3, p 421-437.
- Purkis, S. J., Oehlert, A. M., Dobbelaere, T., Hanert, E., and Harris, P. M., 2023, Always a White Christmas in the Bahamas: temperature and hydrodynamics localize winter mud production on Great Bahama Bank.: Journal of Sedimentary Research, v. 93, p. 145-160.
- Reijmer, J. J. G., Swart, P. K., Bauch, T., Otto, R., Roth, S., and Zechel, S., 2009, A reevaluation of Facies on Great Bahama Bank I: New Facies Maps of Western Great Bahama Bank *in* Swart, P. K., Eberli, G. P., and McKenzie, J. A., eds., Perspectives in Carbonate Geology: A Tribute to the Career of Robert Nathan Ginsburg, IAS Special Publication, Volume 41: Oxford, Wiley-Blackwell, p. 29-46.
- Saenger, C., Affek, H. P., Felis, T., Thiagarajan, N., Lough, J. M., and Holcomb, M., 2012, Carbonate clumped isotope variability in shallow water corals: Temperature dependence and growth-related vital effects: Geochimica et Cosmochimica Acta, v. 99, p. 224-242.
- Shinn, E. A., Steinen, R. P., Lidz, B. H., and Swart, P. K., 1989, Whitings, a sedimentologic dilemma: Journal of Sedimentary Research, v. 59, no. 1, p. 147-161.
- Swart, P. K., Lu, C., Moore, E. W., Smith, M. E., Murray, S. T., and Staudigel, P. T., 2021, A calibration equation between Δ48 values of carbonate and temperature: Rapid Communications in Mass Spectrometry, v. 35, no. 17, p. e9147.
- Swart, P. K., Oehlert, A. M., Mackenzie, G. J., Eberli, G. P., and Reijmer, J. J. G., 2014, The fertilization of the Bahamas by Saharan dust: A trigger for carbonate precipitation?: Geology, v. 42, no. 8, p. 671-674.
- Thiagarajan, N., Adkins, J., and Eiler, J., 2011, Carbonate clumped isotope thermometry of deep-sea corals and implications for vital effects: Geochimica et Cosmochimica Acta, v. 75, no. 16, p. 4416-4425.
- Wang, Z., Schauble, E. A., and Eiler, J. M., 2004, Equilibrium thermodynamics of multiply substituted isotopologues of molecular gases: Geochimica et Cosmochimica Acta, v. 68, no. 23, p. 4779-4797.